

Aminology in Tribology

October 11, 2019 Houston STLE Anabel Rubio, Sr. Scientist Huntsman Performance Products

Topics of Discussion



- Overview of Huntsman Corporation
- Amines for Tribology
 - Introduction
 - Trends
- Amine Basics
- Applications
 - Fuels and Lubricants
 - Metalworking
- EH&S and Regulatory



Huntsman Businesses

Polyurethanes

Performance Products

Advanced Materials

Textile Effects

MDI

Polyols

PO/MTBE

TPU

PU Systems

Amines

Surfactants

Maleic Anhydride

Upstream Intermediates **Composites**

Adhesives

Resins

Dyes

Chemicals

Inks

Apparel
Home & Institutional
Technical Textiles











Amines in Tribology

Tribology is the study of science and engineering of interacting surfaces in relative motion which includes the study and application of friction, lubrication and wear.



Typical Additives in Lubricants

- Antioxidants
- Extreme Pressure/ Antiwear Additives
- Detergents
- Dispersants
- Viscosity Index Improvers
- Friction modifiers/ Boundary Lubricants
- Corrosion Inhibitors
- Metal Deactivators
- Pour Point Depressants
- Foam Inhibitors
- Demulsifers



Global Fuels and Lubricants Development



Market Trends Impacting Amines

Market Trends for Fuels

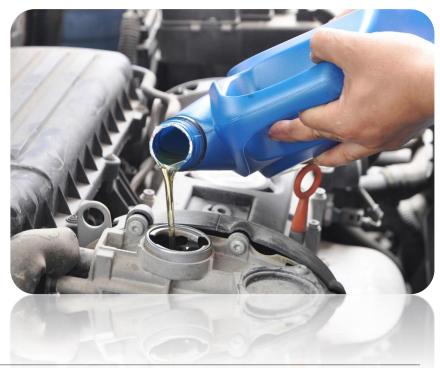
- Higher fuel efficiency standards
- Tighter fuel emission standards
- Move towards direct injection engines and turbocharging to improve fuel economy

Market Trends for Metalworking

- Cleaner labeling
 - Green Additives
- Multi-metal metalworking formulations
- Move to semi-synthetic and synthetic formulations
- Longer fluid life

Market Trends for Lubricants

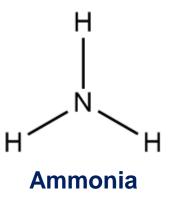
- Reduction in ash limits
- Long drain intervals

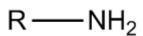




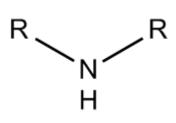
Types of Amines



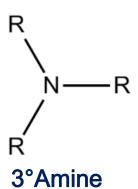


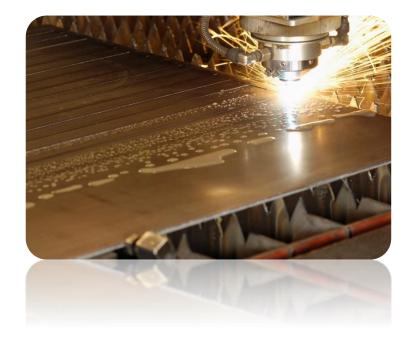


1°Amine



2°Amine





MEA

HUNTSMA Enriching lives through innovation

Examples

DEA

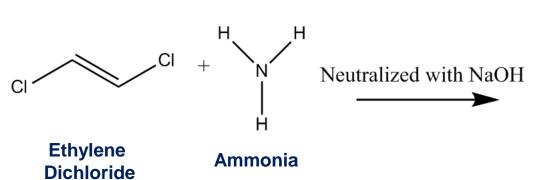
TEA

Ethanolamine Synthesis



Ethyleneamine Synthesis





NaC1



Applications

Ethyleneamines and Polyetheramines

HUNTSMAN Enriching lives through innovation

Use in Lubricants

- Polyetheramines are products used in fuels. For this application, they are usually comprised of an alcohol that is alkoxylated to a degree and aminated. These molecules are balanced to be compatible with the fuel and clean up/ keep clean the combustion chamber
- Ethyleneamines and Polyetheramines can be reacted with PIBSA to create an ashless dispersant for motor oils

Typical dispersant molecules are composed of a polar head (in this case the head is an N group) and a hydrocarbon tail (polybutylene)

 Ethyleneamines and some polyetheramines may be used to create a Mannich base



- Process of working with metals to create individual parts, assemblies, or large-scale structures. This includes a wide range of objects, processes and tools
- Four main processes exist

Removal

Process for which a cutting action is the primary machining technique



Forming

Process that involve changing the shape and contour without cutting



Treating

Process like cleaning or annealing in order to prepare material for the next Process step



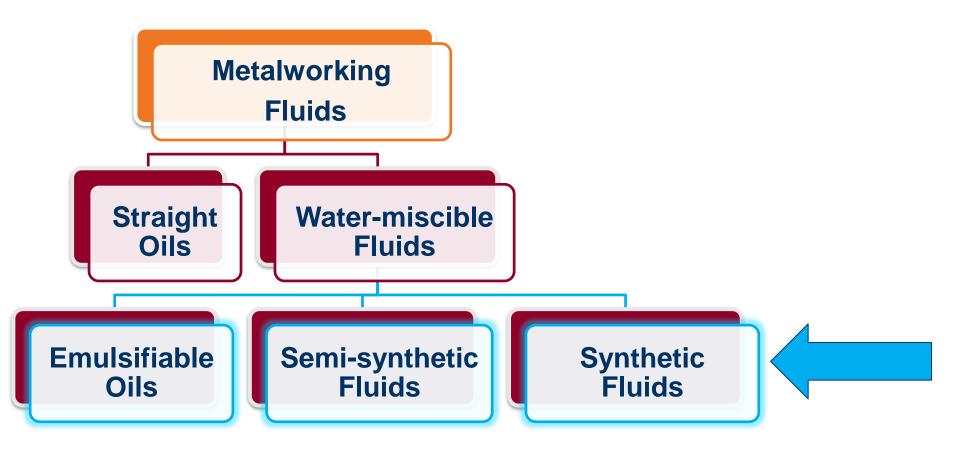
Protecting

Process to preserve metal surfaces from corrosion. This for short or longer term



Fluid types





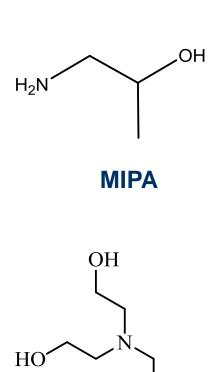


- In water-miscible metalworking formulations, amines help to
 - Maintain dilution pH
 - aids in corrosion protection
 - helps control microbial control
 - higher pH is known to help extend fluid life
- Provide pH reserve alkalinity
 - Neutralize acidic components in formulation to form amine carboxylates
 - Neutralize acidic components that form from oxidation

$$NH_2$$
 H_2N
 OH

MEA

AMP



TEA

OH

Comparison of Physical Properties



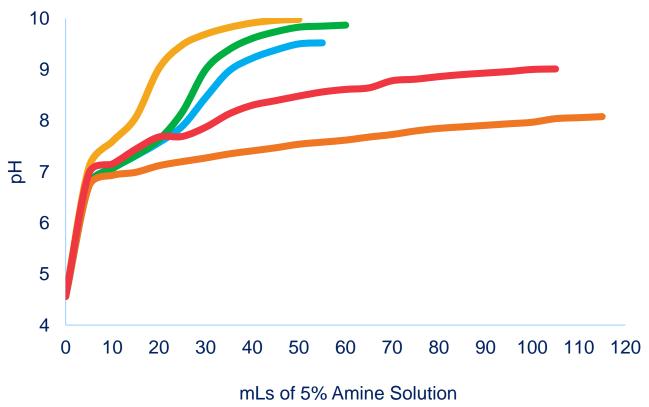
Amine	Molecular Weight	Flash Point ° C	5% pH	рКа
Monoethanolamine (MEA)	61	95	11.87	10.03
Monoisopropanolamine (MIPA)	75	71	11.40	9.40
2-Amino-2-methyl propanol (AMP)	94	78	11.46	9.82
DGA™ Agent	105	128	10.68	9.45
Diethanolamine (DEA)	105	149	10.34	8.88
Triethanolamine (TEA)	149	202	9.64	7.76
Methyldiethanolamine (MDEA)	119	141	10.7	8.76

The higher the pKa the stronger the amine



Titration of 5% Decanoic Acid solution with Different Amines





- MWF can be between 8.5-10.5 pH depending on alloy being used and application
- Primary amines important to attain desired pH
- Tertiary amines important for pH buffering

Other considerations



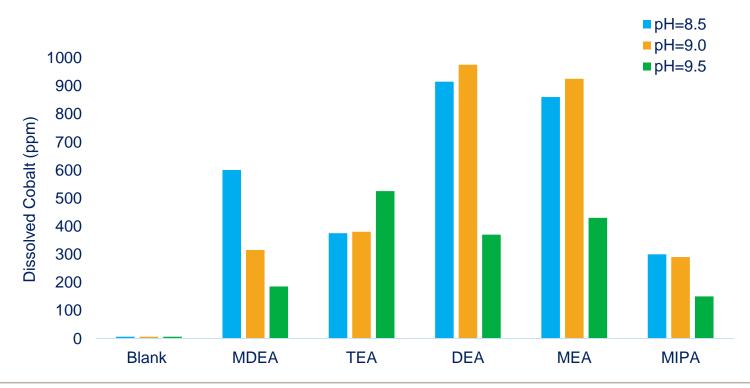
- Odor
- Cobalt Leaching
- Aluminum Staining
- Corrosion inhibition



Cobalt Leaching

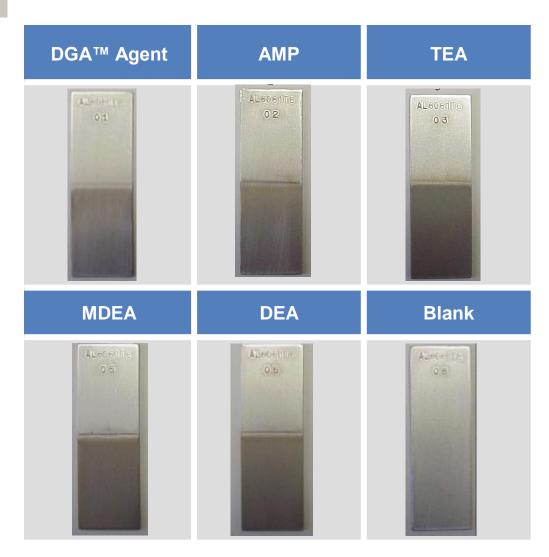


- Cobalt leaching is affected by pH
 - In general lower pH shows higher leaching
- Amine selection is important as certain amines leach cobalt more than others



Aluminum Staining

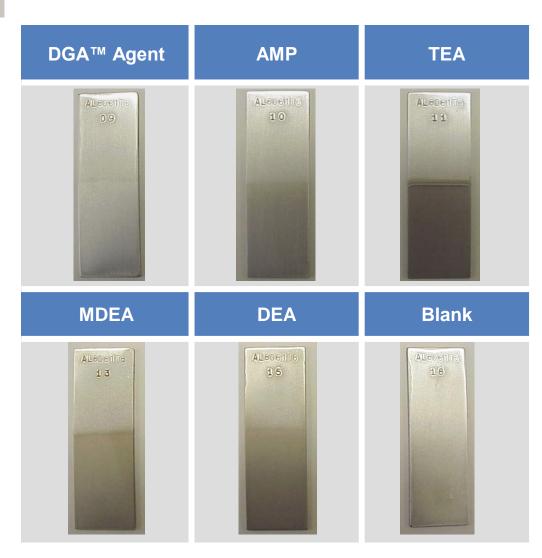




2.5 weight % amine aqueous
Al 6061
24 hours
pH = 8.8

Aluminum Staining





2.5 weight % amine aqueous
Al 6061
24 hours
pH = 8.0

Aluminum Staining





2.5 weight % amineisononanoic acid salt aqueous Al 6061 24 hours pH = 8.8

Aluminum Corrosion



2.5% Amine Aq. Soln.	Dissolved Aluminum				
	AI 7075	AI 6061	AI 2024	AI 390	
DGA™ Agent	5	5.5	7.6	9	
AMP	9	7.9	12	11.5	
DEA	7.9	9.3	10	11	
TEA	155	195	235	215	
MDEA	7.2	4.8	8.6	9	
2.5% Amine Salt Soln. (0.1 mole of amine/0.083 mole of acid)	Dissolved Aluminum				
	AI 7075	Al 6061	AI 2024	AI 390	
DGA™ Agent	3.1	3.8	2.8	1.3	
AMP	4.7	5.5	3.6	4.5	
DEA	2.5	3.8	1.2	2.7	
TEA	40	39	34	32	
MDEA	1.6	2.2	1.4	1.8	

Environmental, Health, and Safety



Desire minimal labeling and skin compatibility

- Amines tend to be corrosive and an irritant
- DEA includes health hazard label for target organ toxicity, repeated exposures



Reduced misting and volatility

Lower vapor pressure preferred for amines

	MEA	DEA	TEA	DGA™ Agent	AMP	MDEA
Exposure limit	6 mg/m ³ OSHA PEL	15 mg/m ³ TWA ACGIH	5 mg/m ³ TWA ACGIH	None	None	None

Regulatory



- Registration on chemical inventories
- TEA and MDEA listed on Chemical Weapons Convention List
- Nitrosamine formation in MWF
 - EPA cautioned on the use of certain amines in combination with nitrites
 - DEA and TEA
- Reduced biocide use
 - EPA under FIFRA has lowered the concentration use of formaldehyde based biocides





Thank you



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